STRUCTURE ANALYSIS BY ¹³C NMR SPECTROSCOPY OF CRIOPHYLLINE, A NEW DIMERIC INDOLE ALKALOID^{(1)*}

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We wish to report on the structure analysis by 13 C nmr of criophylline <u>1</u>, a new dimeric indole alkaloid, $\left[\text{mp } 278-280^{\circ} ; \left[\alpha \right] \right]_{D}^{22} - 176^{\circ}$ (c 1.04 CHCl₃); m/e 646.3528 (calc 646.3519) C₄₀H₄₆N₄O₄, isolated from the leaves of <u>Crioceras dipladeniiflorus</u> (Stapf.) K. Schumann. The elucidation of this highly complex structure is based almost completely on 13 C nmr data⁽²⁾. ¹H nmr and mass spectral analysis provided complementary informations about the constitution of this alkaloid⁽³⁾.

Criophylline does not undergo acetylation under normal conditions. Its 100 MHz pmr spectrum indicates a carbomethoxy function $\left[\delta \ 3.72(s) \right]$, two methyl groups on secondary carbon centers $\left[0.9(t, J = 6.5 \text{ Hz}) \right]$, seven unsaturated hydrogens and several overlapping uninterpretable multiplets. The uv spectrum $\left[\text{EtOH}, \dot{\Lambda} \text{ nm}, (\log \varepsilon) \right]$: 214(4.26), 228(4.06), 257(3.93), 299(4.08), 327(4.28) consists of the addition of two chromophores : one of tabersonine type⁽⁴⁾, the other of dihydroindole type. The mass spectrum $\left[\text{m/e } 432.2650 \left(\text{calc } 432.2651 \text{ C}_{27}\text{H}_{34}\text{O}_2\text{N}_3 \right) \right]$ reveals retro Diels-Alder fragmentation, typical of a tabersonine-like moiety⁽⁴⁾.

Noise and single frequency decoupled 13 C nmr spectra⁽⁵⁾ of criophylline shows below 80 ppm : four nonprotonated carbons, eight methines, ten methylenes and three methyl groups. Above 80 ppm eight quaternary and seven methine carbons can be observed in agreement with the 46 hydrogen atoms of the molecule, suggesting that each of the indole nitrogens bears a hydrogen.

The high number of saturated methine sites (eight) with respect to only one in tabersonine and their relatively high chemical shifts for alco-

^(*) Dedicated to Professor M.-M. Janot on his 70th birthday.

holic oxymethine carbons suggested epoxide linkages for the two unaccounted oxygens. Comparison of the ¹³C nmr data of hazuntinine $2^{(6)}$ and the new dihydroindole alkaloid andrangine $3^{(7)}$ with the spectrum of criophylline afforded proof that the individual components of the dimer were related to these two epoxy-alkaloids.

While each saturated andrangine carbon has its signal unaffected in the 13 C nmr spectrum of criophylline (both from the point of view of chemical shift and single frequency decoupled multiplicity ; see Table and Figure) some hazuntinine carbons below 80 ppm are markedly influenced by the linkage of the two monomers. Inspection of the unsaturated carbon region of the criophylline spectrum indicates an unsubstituted aromatic ring for the indole moiety⁽⁶⁾ while one aromatic protonated carbon of the andrangine part becomes a quaternary site. As a consequence, the latter unsaturated dihydroindole carbon - assigned to C-10^(6,8) - must be attached to a saturated site of the epoxy-tabersonine moiety. As the single frequency decoupled spectrum of criophylline exhibited only three aminomethylene signals, either C-5' or C-3' must be the second site of linkage^{*}.

The lowest field carbon which can be assigned to C-21' appears at 62.0 ppm. Due to the shielding of this signal with respect to C-21' of hazuntinine (70.8 ppm) H-21' must be involved in a strong steric interaction. Either the C-14' - C-15' epoxide in the dimer is on the same α side as H-21' or the linkage of the monomers takes place at the α side at C-5' or C-3', hydrogen-21' being then involved in a 1,3-type diaxial interaction⁽⁹⁾. The almost identical C-17' and C-19' shifts for the monomer and the dimer do not support the first hypothesis.

The ¹³C nmr spectrum of criophylline offers at first sight two different interpretations : in case of a substitution on C-5' α^{**} : (& monomerdimer) C-3' 49.2 - 42.4 (peri interaction) ; C-5' 51.4 - 58.1 (α effect) ; C-6' 43.4 - 48.4 (β effect) ; C-14' 51.8 - 54.2 (?) ; C-15' 57.0 - 56.5 ; C-21' 70.8 - 62.0 (1,3-interaction).

^(*) The prime symbols are applied for convenience to the epoxytabersonine carbons, both in the monomer and the dimer.

^(**) Carbons not discussed have identical shifts in the monomer and in the dimer (Table).

 $\frac{1}{3}$

51.5

51.9

	— —			158.	-TABER	SONINE	4				
	C-2'	C-3'	C-5'	c-6'	C-7'	c-8'	C-9'	C-10'	C-11'	C-12'	C-13'
$\frac{\frac{1}{4}}{\frac{1}{4}}$	165.0 166.0	58.1 49.2	48.4 51.4	42.4 43.4	54.2 54.8	138.0 138.0	121.5 121.0	120.2 120.5	$127.6 \\ 127.4$	109.1 109.3	$143.1 \\ 143.4$
	C-14'	C-15'	C-16'	C-17'	C-18'	C-19'	C-20'	C-21'	C=0 '	OMe'	
<u>1</u> <u>4</u>	56.8 51.8	54.2 57.0	90.8 90.7	23.9 23.3	7.3 7.0	27.0 26.3	36.6 36.8	62.0 70.8	168.8 168.8	50.9 50.7	
	C-2	C-3	C-5	c-6	C-7	c-8	C-9	C-10	C-11	C-12	C-13
$\frac{1}{3}$	75.9 75.3	50.6 ^b 50.4 ^b	52.1 ^b 51.5 ^b	24.2 24.2	44.1 44.2	132.0 131.8	123.9 122.7	125.2 118.9	127.6 127.4	108.6 109.3	149.1 149.5
	C-14	C-15	C-16	C-17	C-18	C-19	C-20	C-21			

TABLE

(a)(b) Assignment may be reversed in horizontal lines.

9.7

9.6

35.0ª

34.9^a

58.3 58.3 34.3ª

 34.2^{a}

(c) Chemical shifts are those observed for hazuntinine $\underline{2}$, for the six aromatic carbons δ values of tabersonine have been given⁽⁶⁾.

35.0ª

35.2^a

43.9

43.8

70.2

70.1



In case of a substitution on C-3' α : C-3' 49.2 -58.1 (α effect) ; C-5' 51.4 - 48.4 (peri interaction); C-6' 43.4 - 42.4 ; C-14' 51.8 - 56.8 (ß effect) ; C-15' 57.0 - 54.2 (1,3-interaction half-chair conformation) ; C-21' 70.8 - 62.0 (1,3-diaxial interaction).

The substitution on C-5' α can be hardly reconciled with the variation of the shift of C-14' and the shielding of C-3' seems to be too strong.

Inspection of the 13 C nmr spectra of a variety of steroidal epoxide models⁽⁹⁾ supports well the 2.8 ppm shielding effect of C-15' upon C-3' α substitution. The agreement between expected and experimental values is much better in this case. While the assignment of the structure is on firm ground, that of the substitution site of the epoxy-tabersonine moiety at C-3' is only tentative at this time.

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